

The Student Volume 5



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Reviews

This publication is definitely not simple to begin on studying but quite fun to see. It really is full of knowledge and wisdom I am just effortlessly can get a satisfaction of studying a created pdf.
(Alfreda Bradtke)

THE STUDENT VOLUME 5



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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 24 pages. Original publisher: Greenbelt, Md. : NASA Goddard Space Flight Center, 2000 OCLC Number: (OCoLC)83756137 Excerpt: . . . 9. 3. 2 Calculated Variables Porewater CH₄ concentration (PWCH₄) was calculated as: CH₄hs P Po Kh Vhs Vw (CH₄hs-CH₄amb) P RT where: CH₄hs is the equilibrated headspace concentration of CH₄ (ppm) P and Po are ambient and sea-level atmospheric pressures (101325 Pa) CH₄ Kh is the Henrys law constant for at ambient temperature (298 K) Vhs and Vw are the volumes of headspace and water in the syringe CH₄amb is the ambient CH₄ concentration (1. 8 ppmv) R is the universal gas constant (8314. 4 L Pa K mol T is the ambient temperature (298 K) Porewater concentration calculated as: (PWCO₂) was CO₂ CO₂hs P Po Kh (I Kal 10pH (I Ka2 10pH)) Vhs Vw (CO₂hs-CO₂amb) P RT where: is the equilibrated headspace concentration of (ppm) CO₂hs CO₂ P and Po are ambient and sea-level atmospheric pressures (101325 Pa) Kh is the Henrys law constant for CO₂ at ambient temperature (298 K) Kal and Ka2 are the 1st and 2rid dissociation constants for aqueous CO₂ pH is the pH of the porewater after stripping Vhs and Vw are the volumes of headspace and water in the syringe CO₂amb is the ambient CO₂ concentration (350 ppmv) R is the universal gas constant (8314. 4 L Pa K tool) T is the ambient temperature (298 K) 9. 4 Graphs and Plots None given. 10. Errors 10. 1 Sources of Error During sampling, a leak in...



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